

2-CYANO  $\Delta^3$  PIPERIDINES III<sup>1</sup> :  
TOTAL SYNTHESIS OF ( $\pm$ ) 20-EPIULEINE

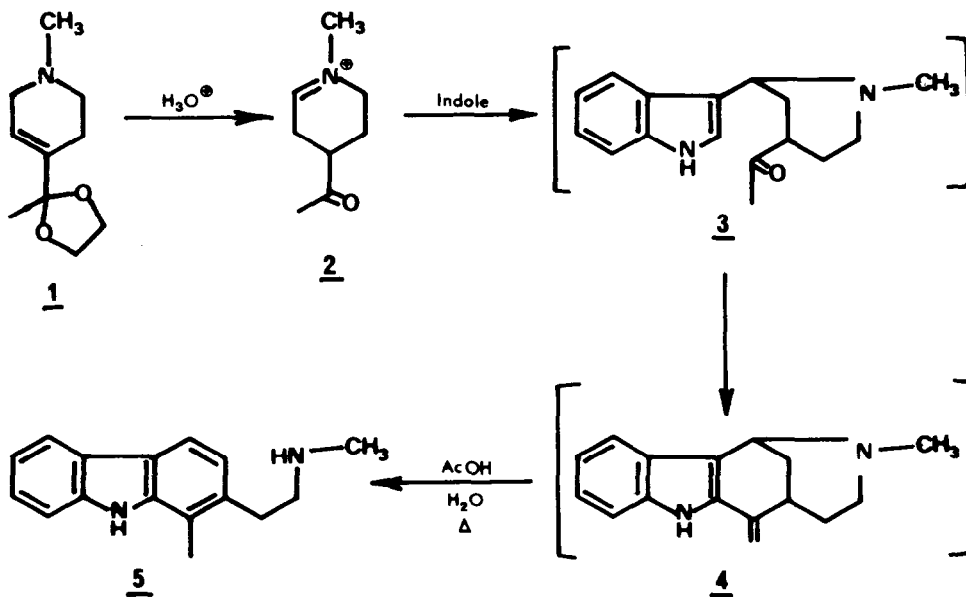
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Summary : The total synthesis of the indole alkaloid ( $\pm$ ) 20-epiuleine has been achieved starting from indole itself and the appropriate 2-cyano  $\Delta^3$  piperidine 7.

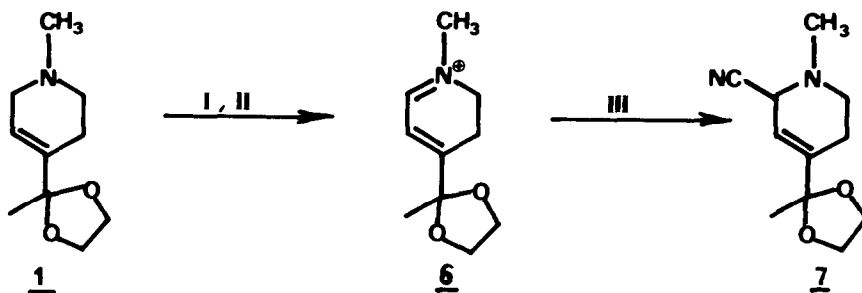
In continuation of our studies on general methods for the synthesis of alkaloids based on 2-cyano  $\Delta^3$  piperidine synthons <sup>1,2</sup> we wish to report a short synthesis of ( $\pm$ ) 20-epiuleine 10<sup>3</sup>.

The strategy employed relates to that of our previous preparation of the carbazole 5 via the postulated intermediacy of 3 and desethyluleine 4<sup>4,5</sup>. In a one pot reaction, condensation of indole with the iminium salt 2 derived from 1, followed by addition-elimination at the keto group, led to desethyluleine which fragmented under the reaction conditions (Scheme 1).



SCHEME 1

For the synthesis in the uleine series, the required intermediate 9 (Scheme 3) analogous to 3 was seen as being derivable by conjugate addition of an ethyl group on the  $\alpha$ ,  $\beta$  - unsaturated ketone 8. The preparation of 8 itself could be envisaged via the condensation of indole with the conjugated iminium salt 6, followed by hydrolysis. The iminium ion 6 represents a 5,6-dihydropyridinium species and we have demonstrated<sup>2</sup> that 2-cyano  $\Delta^3$  piperidines are stable equivalents of this system. Thus we decided to prepare the derivative 7 as a suitable reactive form of the intermediate 6 (Scheme 2).



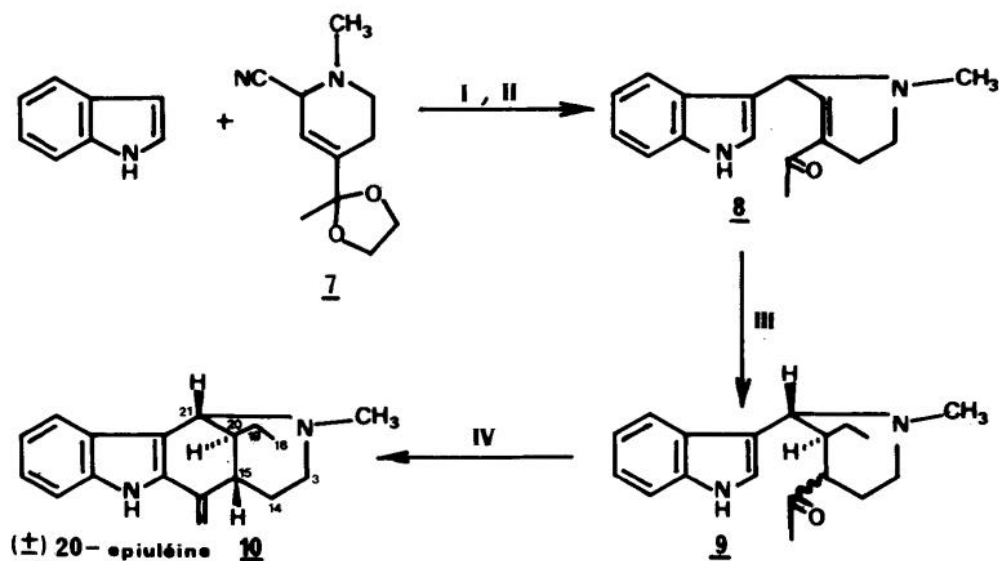
Reagents : I, mCPBA,  $\text{CH}_2\text{Cl}_2$ ,  $20^\circ$ , 30 min. ; II, TFAA,  $\text{CH}_2\text{Cl}_2$ ,  $0^\circ$ , 1h. ; III, KCN,  $\text{H}_2\text{O}$ , pH 4.

SCHEME 2

The required compound 7 was synthesized in high yield from 1<sup>4</sup> via the sequence shown in scheme 2. N-oxidation was effected in 95 % yield on the reaction of 1 with m-chloroperbenzoic acid in  $\text{CH}_2\text{Cl}_2$ . Treatment of the resultant N-oxide with trifluoroacetic anhydride and trapping of the intermediate iminium ion 6 with cyanide led to the formation of 7<sup>6</sup> (89 %). No hydrolysis of the ketal function occurred under the conditions of the reaction (pH4, two phase system<sup>2</sup>). When 7 was stirred with indole in aqueous acetic acid at room temperature for 16 h, after which time aqueous HCl was added to effect ketal hydrolysis, the required ketone 8<sup>7</sup> was obtained in 95 % yield (Scheme 3). Reaction of 8 with EtMgBr and CuCl in THF, afforded 9<sup>8</sup> in 51 % yield. The trans relationship between the protons at C-20 and C-21 of 9 could be ascertained from their axial-axial coupling constant ( $J = 11 \text{ Hz}$ ).

Treatment of 9 in  $\text{CHCl}_3$  with TsOH ( $\Delta$ ,  $\text{N}_2$ , 15 h) gave ( $\pm$ ) 20-epiuleine 10 in 56 % yield. This last step is the same as that carried out by Natsume and Kitagawa<sup>9,10</sup> in their recent synthesis of the same alkaloid.

The described synthesis of ( $\pm$ ) 20-epiuleine is a further example of the utility of 2-cyano  $\Delta^3$  piperidines in the total synthesis of alkaloids.



Reagents : I, AcOH-H<sub>2</sub>O (1:1), rt, 16h. ; II, HCl 10% ; III, EtMgBr, CuCl, THF, 0°, 5 min. ; IV, TsOH, CH<sub>2</sub>Cl<sub>2</sub>, Δ, N<sub>2</sub>, 15h.

SCHEME 3

#### AKNOWLEDGEMENTS :

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#### REFERENCES AND NOTES

1. This work was presented for the first time in the course of a lecture at the 12th International Symposium on the Chemistry of Natural Products, September 1980.  
Part II see : M. HARRIS, D.S. GRIERSON, C. RICHE and H.-P. HUSSON, Tetrahedron Lett., 1980, 1957-1960.
2. D.S. GRIERSON, M. HARRIS and H.-P. HUSSON, J. Am. Chem. Soc., 1980, 102, 1064-1082.
3. For the other syntheses in the uleine series, see ref. 9.
4. R. BESSELIÈVRE and H.-P. HUSSON, Tetrahedron Lett., 1976, 1873-1876.

5. R. BESSELIÈVRE and H.-P. HUSSON, Tetrahedron, Woodward Memorial Issue, in press.
6. 7 : liquid ; MS m/e (relative intensity) :  $M^{+}$  208 (10), 180 (15), 121 (17), 87 (100);  $^1H$  NMR ( $CDCl_3$ , 60 MHz, TMS,  $\delta = 0$ ) 1.43 (3H, s,  $CH_3$ ), 2.49 (3H, s,  $NCH_3$ ), 3.90 (4H, m,  $O(CH_2)_2O$ ), 4.1 (1H, m,  $CH$  CN), 5.9 (1H, m,  $CH = C$ ).  
All yields refer to pure chromatographed products.
7. 8 : amorphous ; MS m/e (relative intensity) :  $M^{+}$  254 (5), 211 (100), 168 (40), 43 (60) ;  $^1H$  NMR ( $CDCl_3$ , 60 MHz, TMS,  $\delta = 0$ ) 2.20 (3H, s,  $CH_3CO$ ), 2.25 (3H, s,  $NCH_3$ ), 4.25 (1H, m, C-21  $H$ ), 6.80 (1H, m,  $CH = C$ )
8. 9 : amorphous ; MS m/e (relative intensity) :  $M^{+}$  284 (30), 241 (100), 198 (20), 130 (20)  $M^{+}162.67$  (241  $\rightarrow$  198) ;  $^1H$  NMR ( $CDCl_3$ , 400 MHz, TMS,  $\delta = 0$ ), 0.70 (3H, t,  $J = 6Hz$ ), 2.00 (3H, s,  $CH_3CO$ ), 2.18 (3H, s,  $NCH_3$ ), 3.05 (1H, d,  $J = 11Hz$ , C-21  $H$ ).  
The major byproduct (32 %) was the secondary alcohol coming from 1, 2 addition to the  $\alpha$ ,  $\beta$  -unsaturated ketone.
9. M. NATSUME and Y. KITAGAWA, Tetrahedron Lett., 1980, 839-840.
10. A yield of 36 % for this step is reported by the authors<sup>9</sup> (reagent quantities not specified). We employed 100 ml  $CHCl_3$ , 700 Mg TsOH for 1 mmole of 9.

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